



University of Baghdad  
College of Eng.  
Dept. of Civil Eng.

Final Exam. 1st Attempt  
Environmental & Sanitary Eng.  
**Answer Six questions only**

Date 8 /June/2013  
Time 3 hrs.  
Forth Year

**Q1)** The chemical analysis of the treated water (potable) from a WTP is given:

Ca (mg/l)	Mg (mg/l)	Al (mg/l)	Fe (mg/l)	Mn (mg/l)	K (mg/l)	Alkalinity mg/l as CaCO <sub>3</sub>
55	23	0.11	0.05	20	5.5	136
HCO <sub>3</sub> (mg/l)	CO <sub>3</sub> (mg/l)	SO <sub>4</sub> (mg/l)	Cl(mg/l)	PO <sub>4</sub> (mg/l)	Na	Conductivity (μmohs/cm)
120	24	120	50	0.02	15.1	516

For the raw water total solids = 395 mg/l, SS = 45 mg/l and conductivity = 510 μmohs/cm (assume f is the same for the treated and raw water). **FIND 1- TDS mg/l** in the treated water **2- Carbonate** and noncarbonated hardness mg/l as CaCO<sub>3</sub> for the treated water. **3- Volume** of a zeolite filter (m<sup>3</sup>) to reduce the effluent hardness to 150 mg/l as CaCO<sub>3</sub> in a drinking water system. The zeolite capacity is 10 kg/m<sup>3</sup>, influent flow 250 m<sup>3</sup>/d.

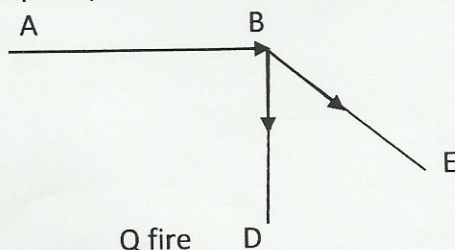
Equivalent weights

Ca 20	Mg 12	Na 23	K 39	Fe 26	Mn 27.5	Zn 33	Al 9
HCO <sub>3</sub> 61	SO <sub>4</sub> 48	Cl 45.5	CO <sub>3</sub> 30	PO <sub>4</sub> 31.7	NO <sub>3</sub> 62	NO <sub>2</sub> 46	F 19

**Q2)** A WTP is to treat 500 m<sup>3</sup>/hr raw water of 150 mg/l SS concentration and 100 NTU turbidity, knowing that  $G_s = 1.2$ ,  $\rho_w = 10^3 \text{ kg/m}^3$  and  $\mu_w = 1.027 \times 10^{-3} \text{ kg/m.s}$ . The circular clarifier was designed according to the column analysis given below, at 40 min settling time. **FIND 1-Diameter** (mm) of the smallest particle to be 100% removed. **2-Volume** of the sludge hopper (m<sup>3</sup>) if the tank is to be cleaned twice a day (assume solid content of the sludge is 5%).

Height (m)	0	0.5	1.0	1.5	2.0
% removal	100	88	73	69	64

**Q3)** For the water distribution system given below, the velocity in pipe BE is 0.8 m/sec. At D a 4 story ordinary type building was constructed, each floor is 100 m<sup>2</sup> and 3 m in height.  
**1- FIND Q** fire demand (m<sup>3</sup>/sec) at D. **2- WHAT** is the minimum pressure at A (KPa) to provide the adequate pressure for fire demand at D. (KPa= 0.102 m)



Line	Length (m)	Dia (mm)	C
AB	650	300	140
BE	300	200	120
BD	500	250	120

- Q4)** 1- State 4 main differences (operation parameters) between Trickling filters and Activated Sludge processes as biological units.  
2-What are the main differences between the nephelometric and photometric methods in water testing.  
3-What is the best type of residual chlorine in potable water. Why?  
4-What are the objectives of: Skimming tanks, Digesters, Carbonation in RSFs, Catch basin in sewer systems.  
5-Explain why it is hard to reach ideal settling in sedimentation tanks.



Q5) A waste water treatment plant is to receive  $500 \text{ m}^3/\text{hr}$  flow with  $\text{BOD}_u = 420 \text{ mg/l}$  and  $300 \text{ mg/l}$  suspended solid concentration. In the primary treatment; removal of 70% for suspended solids and 30% of BOD is achieved. The secondary treatment is to achieve an effluent of 30/30.

1)-If an activated sludge process is used:  $\text{MLSS} = 2300 \text{ mg/l}$ , under flow SS concentration  $= 15 \times 10^3 \text{ mg/l}$ , sludge age  $= 10$  days, decay coefficient  $= 0.05/\text{day}$ , yield coefficient  $= 0.6$ ,  $\text{BOD}_5$  of SS effluent is 65% of the effluent SS concentration and the solids are 80% volatile.

2)-If a single stage trickling filter is used: recirculation ratio  $= 165\%$  and depth of the filter  $= 3 \text{ m}$ .

Compare between the sizes ( $\text{m}^3$ ) of these units and their organic loading rates ( $\text{Kg}/\text{m}^3/\text{d}$ ).

Q6) A river of  $0.75 \text{ m}^3/\text{sec}$  flow with BOD ultimate  $= 3.3 \text{ mg/l}$  and  $\text{DO} = 9.17 \text{ mg/l}$  at  $20^\circ\text{C}$  receives a waste effluent of  $0.25 \text{ m}^3/\text{sec}$  with  $\text{BOD}_{1/30} = 18.3 \text{ mg/l}$  and  $\text{DO} = \text{zero}$  at  $20^\circ\text{C}$ . The flowing velocity in the river is  $0.2 \text{ m/sec}$  with an aeration rate of  $0.82/\text{day}$  and deoxygenating rate  $0.43/\text{day}$  at  $20^\circ\text{C}$ . A position  $50 \text{ km}$  downstream is to be chosen for an intake location, **FIND 1- DO (mg/l). 2-Concentration (mg/l)** of the remaining organic matter.

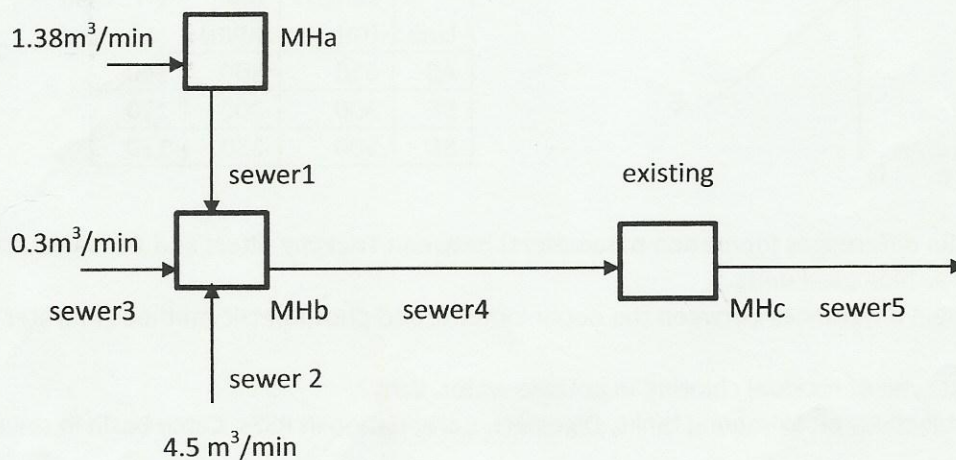
Dissolved Oxygen Saturation Levels							
Temp. $^\circ\text{C}$	0	5	10	15	20	25	30
DO mg/l	14.6	12.8	11.33	10.15	9.17	8.38	7.66

Q7) For the given sewer system, invert to invert at the manholes  $= 25 \text{ mm}$ , **V full** in sewer 1  $= 0.7 \text{ m/sec}$ . Standard sizes (mm) 200, 250, 380, 460, 530, 610, 690, and 760 (assume  $n = 0.013$ ).

1-FIND the standard size for sewer 4 (mm) laid on slope  $0.006$  and  $d/D = 0.7$ . 2- WHAT is the invert depth for sewer 4 at MHb & MHc?

Sewer	Length (m)	Dia. (mm)	Depth of invert (m)
1	40	250	at MHa 0.7 & MHb?
2	45	380	at MHb 0.713
3	50	250	at MHb 0.75
4	50	?	at MHb? & MHc?
5	50	760	at MHc 2.3

Partial Flow Elements					
d/D	q/Q	V/V	d/D	q/Q	V/V
0.3	0.2	0.77	0.6	0.67	1.05
0.4	0.33	0.9	0.7	0.82	1.10
0.5	0.5	1.0	0.8	0.89	1.14



$$Pt = Po + K\Delta t$$

$$F = 18 C (A)^{0.5}$$

$$G = 1020 (P)^{0.5} \{1 - (P)^{0.5} / 100\}$$

$$P = C_D \rho_w \sum A v d^3 / 2$$

$$G = (P / vol \mu)^{0.5}$$

$$y = 1.73 ( \sqrt[3]{qt^2 / (g b^2)} )$$

$$BOD_5 = L (1 - e^{-kt})$$

$$K_{1(T)} = K_{1(20)} 1.047^{T-20}$$

$$K_{2(T)} = K_{2(20)} 1.025^{T-20}$$

$$(C_i - C_e) / C_i = 1 / \{1 + 0.532 (Q C_i / VF)^{0.5}\}$$

$$F = (1+r) / (1 + 0.1 r)^2$$

$$XV = YQ(S_o - S) \theta_c / (1 + k_d \theta_c)$$

$$Q_w = \text{total solid production} / X_r$$

$$Q_r = Q X / (X_r - X)$$

$$\text{Organic loading} = BOD_u Q / Vol$$

$$\ln(Pt) = \ln(Po) + K\Delta t$$

$$V_s = \{4 g (G_s - 1) d / (3 C_D)\}^{0.5}$$

$$V_s = g \rho_w (G_s - 1) d^2 / (18 \mu)$$

$$V_h = \{8 \beta g (G_s - 1) d / f\}^{0.5}$$

$$V = 0.85 C R^{0.63} S^{0.54}$$

$$Q = 0.278 C D^{2.63} S^{0.54}$$

$$V = 0.395 D^{0.67} S^{0.5} / n$$

$$V = 30.385 D^{0.67} S^{0.5}$$

$$Q = 0.31 D^{2.67} S^{0.5} / n$$

$$Q = 23.846 D^{2.67} S^{0.5}$$

$$D = [(e^{-K_1 t} - e^{-K_2 t}) K_1 L / (K_2 - K_1)] + D_i e^{-K_2 t}$$

$$D_c = K_1 (L e^{-K_1 t_c}) / K_2$$

$$t_c = [1 / (K_2 - K_1)] \ln [ (K_2 / K_1) (1 - \{D_i (K_2 - K_1) / L K_1\}) ]$$

$$O_2 \text{ demand} = 1.47 (S_o - S) Q - 1.14 X_r Q_w$$

$$\text{Hydraulic loading} = (Q + Q_r) / A_s$$

التوفيق من الله



Q1 TDS in raw =  $395 - 45 = 350 \text{ mg/l}$   
 $f \text{ in raw} = \frac{\text{TDS}}{\text{Cond}} = \frac{350}{510} = 0.686$

$\therefore \text{TDS in the treated water} = 0.686 \times 510 = 353.98$   
 $= 354 \text{ mg/l}$

T. Hardness =  $\frac{\text{Ca} \times 50}{20} + \frac{\text{Mg} \times 50}{12} + \frac{\text{Al} \times 50}{9} + \frac{\text{Fe} \times 50}{26}$   
 $+ \frac{\text{Mn} \times 50}{27.5}$

$= 137.5 + 95.8 + 0.6 + 0.1 + 36.4$   
 $= 270.4$

ALK = 136  $\text{mg/l as CaCO}_3$  T.H > T. ALK

$\therefore \text{Carb Hardness} = 136 \text{ mg/l as CaCO}_3$

$\frac{136}{136} = 1 \text{ 34.4 mg/l as CaCO}_3$

270.4



150

$Q_p \times 270.4 = 250 \frac{\text{m}^3}{\text{d}} \times 150$

$\therefore Q_p = 138.68 \text{ m}^3/\text{d}$

$Q_z = 111.32 \text{ m}^3/\text{d}$

loading rate =  $111.32 \frac{\text{m}^3}{\text{d}} \times 270.4 \frac{\text{mg}}{\text{l}} \times \frac{1}{1000} = 30.1 \text{ kg/d}$

$\text{Vol} = \frac{30.1}{10} = 3.0 \text{ m}^3$

$\frac{96 \times 61}{50} = \frac{40 \times 30}{50}$

117 24

$$Q_2 \quad v_s = \frac{g \cdot \rho (G_s - 1) d^2}{18 \mu} = \text{SOR}$$

$$\text{SOR} = \frac{H}{t \times 1.5} = \frac{2}{40 \times 1.5} = 3.33 \times 10^{-2} \text{ m/min}$$

$$5.56 \times 10^{-4} = \frac{9.81 \times 10^3 (1.2 - 1) d^2}{18 \times 1027 \times 10^3}$$

$$5.56 \times 10^{-4} = \frac{9.81 \times 10^3 (1.2 - 1) d^2}{18 \times 1027 \times 10^3}$$

$$d = 7.2 \times 10^{-2} \text{ mm} (0.072)$$

$$\text{overall removal} = \frac{0.5}{2 \times 2} (100 + 2 \times 88 + 2 \times 73 + 2 \times 69 + 64)$$

$$= \frac{0.5}{2 \times 2} (100 + 176 + 146 + 138 + 64)$$

$$= 77.9\% \approx 78\%$$

$$\text{Settled Solids} = 500 \frac{\text{m}^3}{\text{hr}} \times \frac{150 \text{ kg}}{1} \times 0.78 \times \frac{10^3}{10^6} = 58.5 \text{ kg/hr}$$

$$= 14.04 \text{ kg/d}$$

$$0.05 = \frac{1404}{1404 + ww}$$

$$0.05 \times 1404 + 0.05 ww = 1404$$

$$ww = 26676 \text{ kg}$$

$$\text{Sludge vol} = \frac{1404}{1.2 \times 10^3} + \frac{26676}{10^3} = 27.9 \text{ m}^3$$

$$\approx 28 \text{ m}^3$$

$$\text{Vol of hopper} = \frac{28}{2} = 14 \text{ m}^3$$



Q5 To Sec. treatment-

$$0.3 = \frac{420 - C_e}{420}$$

$$C_e = 420 \times 0.7 = 294 \text{ mg/l BOD}_u$$

$$\text{BOD}_5 = 294 \times 0.58 = 199.92 \text{ mg/l} \approx 200 \text{ mg/l}$$

1) A.S.

$$S = 30 - 0.65 \times 30 = 10.5 \text{ mg/l}$$

$$XV = \frac{0.6 \times 500(200 - 10.5) \times 10}{1 + 0.05 \times 10} \quad \frac{\text{M}}{\text{hr}} \times \frac{\text{mg}}{\text{l}} \times \frac{10^3}{1} \times 24$$

$$= 9096 \times 10^6 \text{ mg}$$

$$\therefore V = \frac{9096 \times 10^6 \text{ mg} \times \frac{1}{\text{mg}}}{2300 \times 10^3}$$

$$= 3954.78 \approx 3955 \text{ m}^3$$

$$\text{org. loading} = \frac{294 \times 500}{3955} \quad \frac{\text{mg} \times \text{m}^3}{\text{l} \times \text{hr}} \times \frac{10^3}{\text{m}^3 \times 10^6} \times \frac{24}{\text{d}}$$

$$= 0.89 \text{ kg/m}^3/\text{d}$$

2) T.F.

$$F = \frac{1+r}{(1+0.1r)^2} = \frac{1+1.65}{(1+0.1 \times 1.65)^2} = 1.953$$

$$Q = 500 \frac{\text{m}^3}{\text{hr}} \times \frac{1}{60} = 8.33 \text{ m}^3/\text{min}$$

$$\frac{200 - 30}{200} = 0.85 = \frac{1}{1 + 0.532 \sqrt{\frac{8.33 \times 200}{V \times 1.953}}}$$

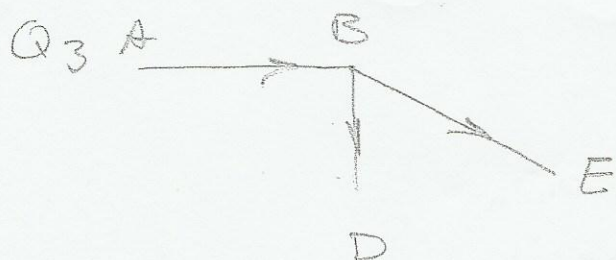
$$1 = 0.85 + \frac{13.207}{\sqrt{V}}$$

$$0.15 = \frac{13.207}{\sqrt{V}}$$

$$\therefore V = 7752.2 \approx 7752 \text{ m}^3$$

$$\text{org. loading} = \frac{294 \times 500}{7752} \times \frac{24}{10^3} = 0.46 \text{ kg/m}^3/\text{d}$$

$\therefore$  A.S is small in size and can receive twice the org. load



$$\begin{aligned}
 1- \text{ at D Fire demand} &= 180 \sqrt{A} \\
 &= 18 \times 1 \times \sqrt{4 \times 100 \times 10.76} \\
 &= 1181 \text{ gpm} \times \frac{3.78}{10^3} \\
 &= 4.46 \frac{\text{m}^3}{\text{min}} \times \frac{1}{60} \\
 &= 7.4 \times 10^{-2} \text{ m}^3/\text{sec}
 \end{aligned}$$

$$\begin{aligned}
 2- Q_{BE} &= \frac{\pi D^2}{4} \times V = \frac{\pi (0.2)^2}{4} \times 0.8 \\
 &= 2.5 \times 10^{-2} \text{ m}^3/\text{sec} \\
 &= 1.5 \text{ m}^3/\text{min}
 \end{aligned}$$

$$\begin{aligned}
 Q_{AB} &= Q_{BE} + Q_{BD} \\
 &= 2.5 \times 10^{-2} + 7.4 \times 10^{-2} \\
 &= 0.099 \text{ m}^3/\text{sec}
 \end{aligned}$$

$$Q = 0.278 C (D)^{2.63} (S)^{0.54}$$

$$Q_{AB} = 0.099 = 0.278 \times 140 (0.3)^{2.63} (S)^{0.54}$$

$$S = 5.5 \times 10^{-3} \therefore h_d = 3.6 \text{ m}$$

$$Q_{BD} = 7.4 \times 10^{-2} = 0.278 \times 120 (0.25)^{2.63} (S)^{0.54}$$

$$S = 1.05 \times 10^{-2} \therefore h_d = 5.2 \text{ m}$$

$$h_{ABD} = 3.6 + 5.2 = 8.8 \text{ m}$$

$$\text{Pressure at D} = 4 \times 3 = 12 \text{ m}$$

The min Pressure at A

$$12 + 8.8 = 20.8 \text{ m}$$

$$\frac{20.8}{0.102} = 204 \text{ kPa}$$



Q7

$$Q_3 = (1.38 + 0.3 + 4.5) \times \frac{1}{60} = 0.103 \text{ m}^3/\text{sec}$$

sewer 4  $\frac{d}{D} = 0.7 \therefore \frac{Q}{Q} = 0.82$

$$\frac{1}{2.67} = 0.375$$

$$\therefore Q = 0.126 \text{ m}^3/\text{sec}$$

$$0.126 = 23.846 (D)^{2.67} (0.006)^{0.5}$$

$$D = 0.386 \text{ m} = 386 \text{ mm}$$

check 250, 380

$$Q_{250} = 23.846 (0.25)^{2.67} (0.006)^{0.5} = 0.046 \text{ m}^3/\text{sec}$$

$$\frac{Q}{Q} = \frac{0.103}{0.046} > 1 \text{ over flow}$$

$$Q_{380} = 23.846 (0.38)^{2.67} (0.006)^{0.5} = 0.139 \text{ m}^3/\text{sec}$$

$$\frac{Q}{Q} = \frac{0.103}{0.139} = 0.74 \therefore D = 380 \text{ mm}$$

sewer 1 invert at MHa = 0.7

$$0.7 = 30.385 (0.25)^{2.67} (S)^{0.5}$$

$$S = 0.0034$$

$$d = 0.0034 \times 40 = 0.136 \text{ m}$$

$$\text{invert at MHb} = 0.7 + 0.136 = 0.836 \text{ m}$$

sewer 2 at MHb = 0.713

sewer 3 at MHb = 0.75

$$\therefore \text{sewer 4 at MHb} = 0.836 + 0.025 = 0.861 \text{ m}$$

$$d = 0.006 \times 50 = 0.3 \text{ m}$$

$$\text{sewer 4 at MHC} = 0.861 + 0.3 = 1.161 \text{ m}$$

sewer 5 at MHC invert 2.3

$$\text{crown } 2-3 - 0.76 = 1.54$$

sewer 4  
1.161  
1.54



$$Q_{mix} = 0.75 + 0.25 = 1 \text{ m}^3/\text{sec}$$

$$DO_{mix} = 0.75 \times 9.17 + 0.25 \times 2.00 = 6.878 \text{ mg/l}$$

$$BOD_{1/30} = BOD_u (1 - e^{-k_1(30)}) \times 1$$

$$k_1(30) = k_1(20) (1.047)^{30-20}$$

$$= 0.43 \times 1.047^{10} = 0.68 \text{ /d}$$

$$18.3 = BOD_u (1 - e^{-0.68 \times 1})$$

$$BOD_u = 37.1 \text{ mg/l}$$

$$\therefore BOD_u/mix = 0.75 \times 18.3 + 0.25 \times 37.1 = 11.75 \text{ mg/l}$$

$$T_{mix} = 20^\circ \text{C} \therefore DO_{sat} = 9.17 \text{ mg/l}$$

$$D_i = 9.17 - 6.878 = 2.292 \text{ mg/l}$$

$$t = \frac{\text{dist}}{V} = \frac{50 \times 10^3}{0.2 \times 24 \times 3600} = 2.894 \text{ day}$$

$$D = \frac{0.43 \times 11.75}{0.82 - 0.43} (e^{-0.43 \times 2.894} - e^{-0.82 \times 2.894}) + 2.292 e^{-0.82 \times 2.894}$$

$$= 2.74 \text{ mg/l}$$

$$\therefore DO \text{ at } 50 \text{ km} = 9.17 - 2.74 = 6.43 \text{ mg/l}$$

$$y \text{ at } 50 \text{ km} = L e^{-k_1 t}$$

$$\text{remaining organic matter} = 11.75 e^{-0.43 \times 2.894} = 3.39 \text{ mg/l}$$